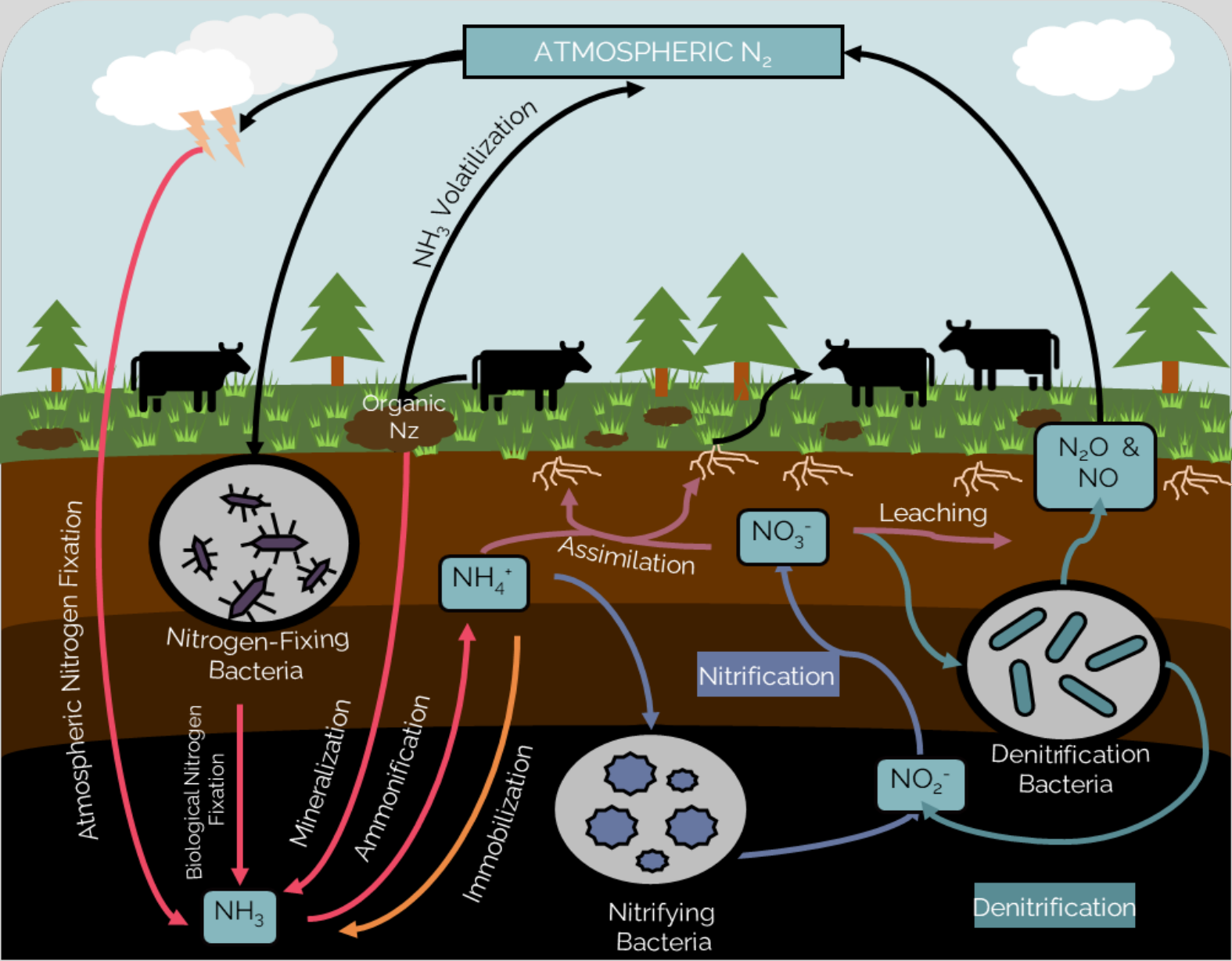


The integration of manure, soil, diet and animal management to reduce on farm GHG

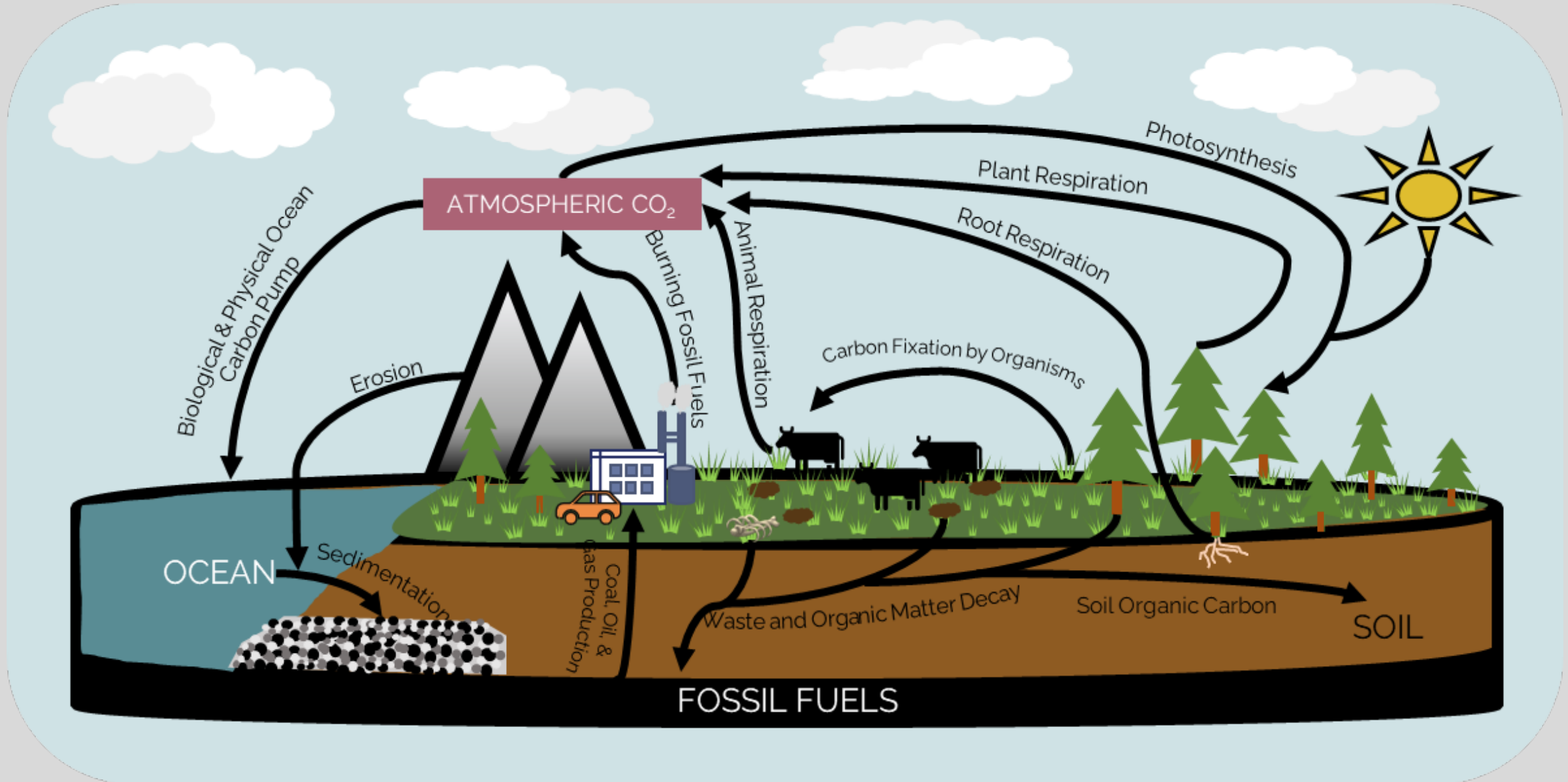
Kristen Johnson
Washington State University
Pullman, WA USA



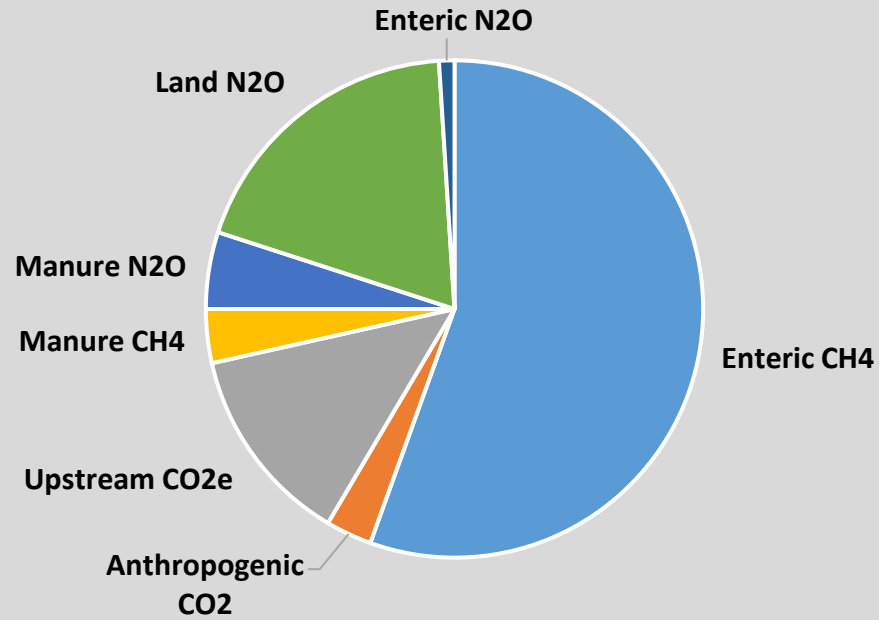
NITROGEN CYCLE



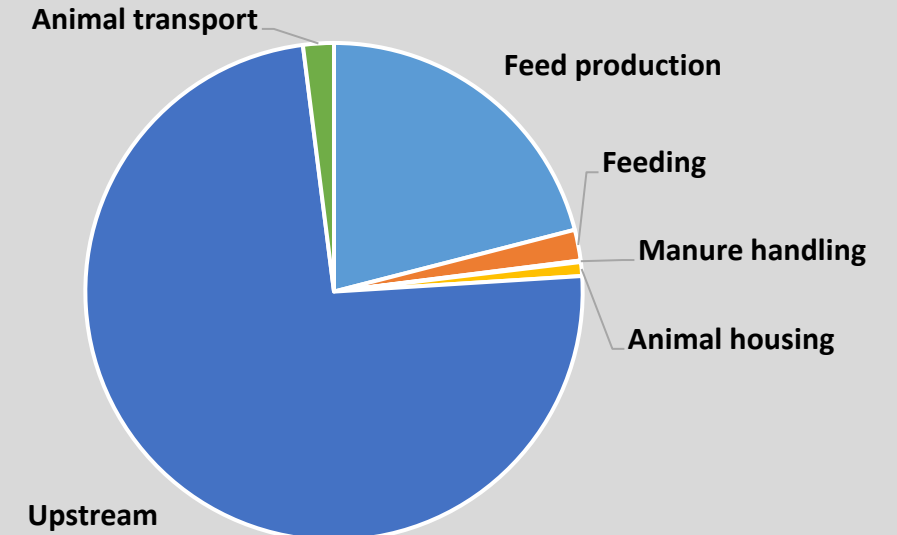
CARBON CYCLE



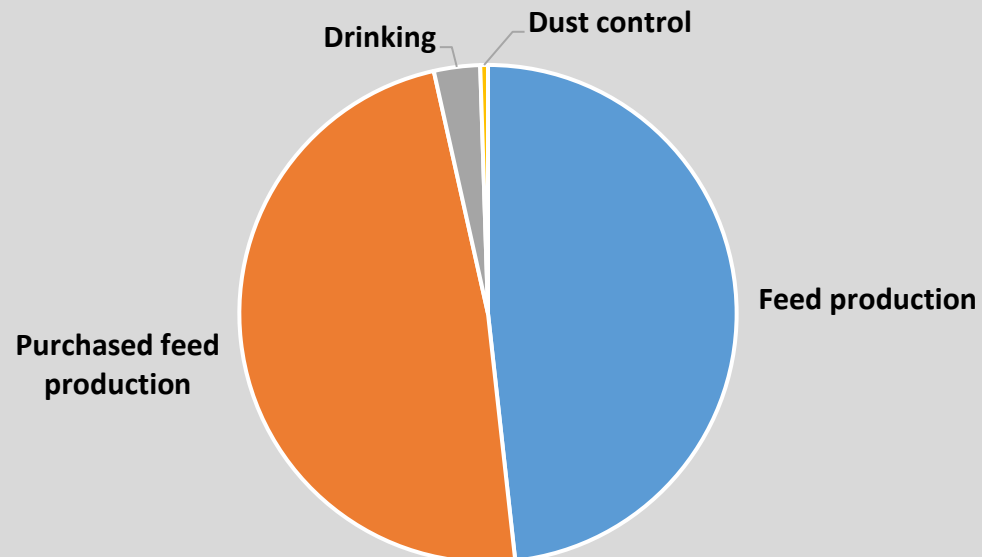
Greenhouse Gas Emission



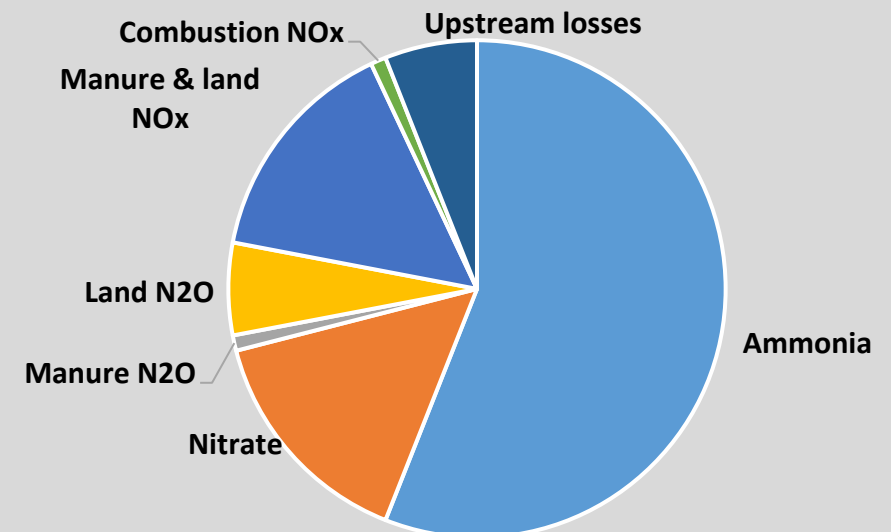
Energy Use



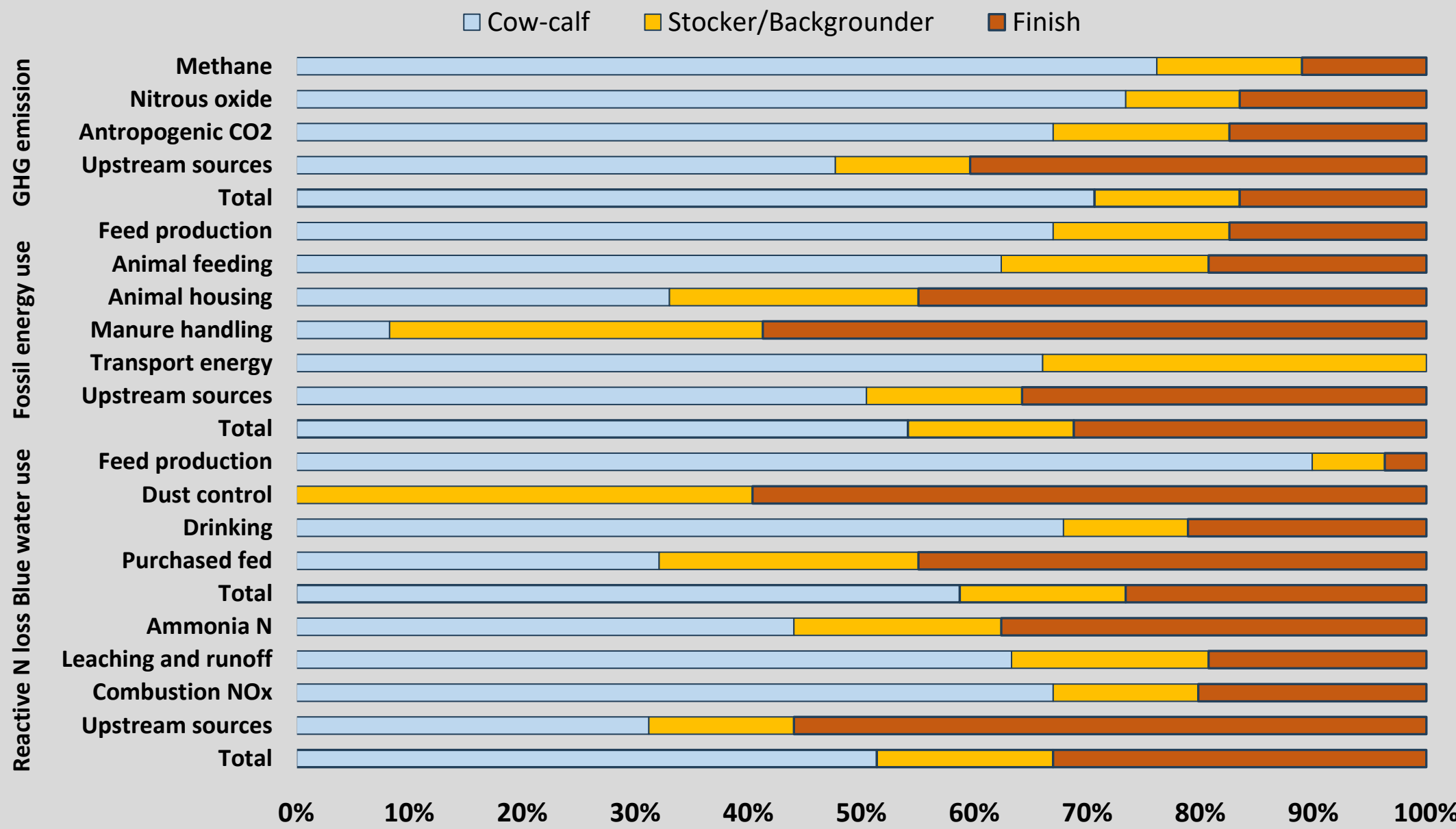
Blue Water Use



Reactive Nitrogen Loss



Distribution of environmental impact sources across the three major phases of the beef cattle production life cycle



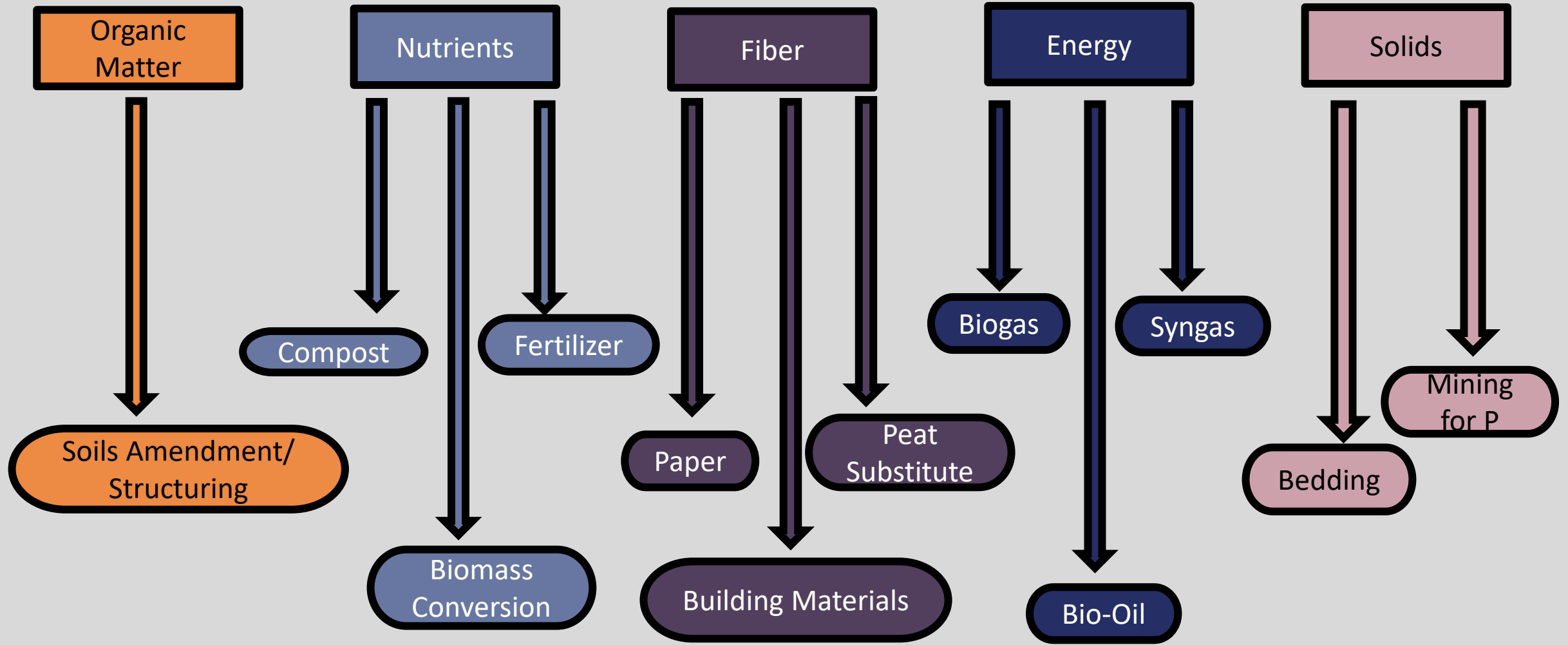
Reduction in enteric CH₄ emissions

- Improved digestibility of feed
- Feed additives
- Improvement in productivity/efficiency
- And others....many papers at this meeting

But there are trade offs.....

- Manure emissions changed (e.g. digestors etc)
- Soil microbial population changes?
- Changes in types of fertilizer used (e.g. LCA)

MANURE BENEFITS



ENVIRONMENTAL CONCERNS FROM MANURE

Anaerobic decomposition of manure increases odorous compounds,

(Pfoest et al., 2018)

Manure accounts for 14 % of GHG emission within the agriculture sector in the US (EPA, 2021)

20-45 % of the N in a lagoon lost as NH_3 (Rotz and Leytem, 2015)

P and K build up in lagoon sludge

(Sharara and Kulesza, 2020)



Photo courtesy of Jael Batty at Probiotic Solutions

TYPES OF MANURE HANDLING ON DAIRIES

Aerobic Handling

Composting



Photo courtesy of WSU

Vermifiltration



Anaerobic Handling

Anaerobic Digestion

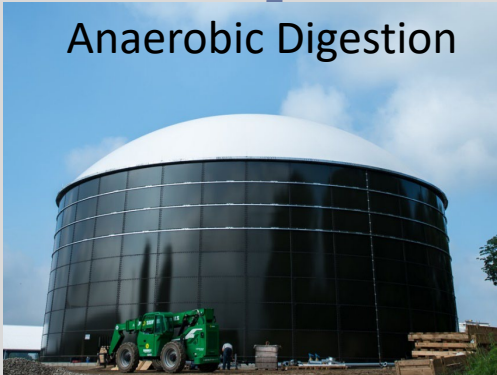


Photo courtesy of Michigan State University

Lagoon



Photo courtesy of UC-Davis

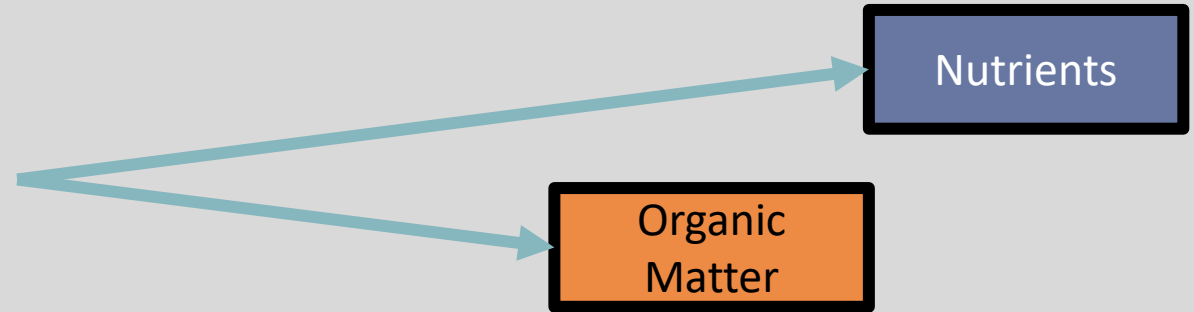
Static Pile



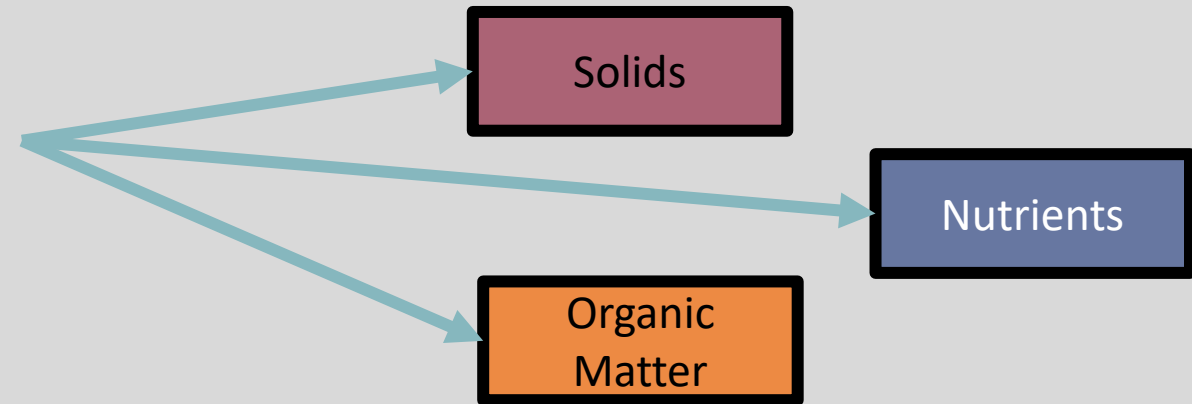
Photo courtesy of Lawn Care

POTENTIAL VERMIFILTRATION BENEFITS

Cleaner
wastewater/vermitea



Vermipost
production



Reduced
Emissions

? ? ?

Creating a soil management bank

PLANT CARBONS

Carbon from plants is transferred to the soil via roots and decomposition, and flows into grazing animals.

RESPIRATION
Carbon dioxide (CO₂) released by plants and animals.

PHOTO-SYNTHESIS
CO₂ captured by plants.

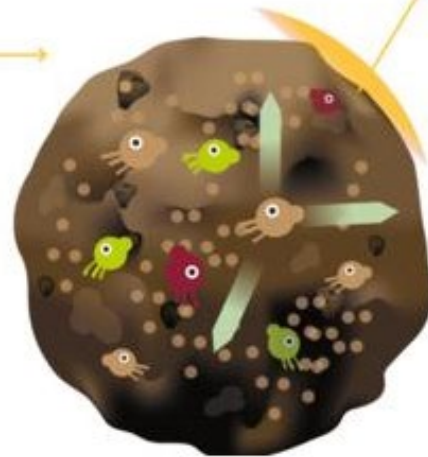
DECOMPOSITION AND MANURE
Carbon released by animals.

SOIL CARBON

SOIL ORGANISMS

Soil organisms are constantly processing animal and plant waste and releasing beneficial nutrients and pure carbon into the soil.

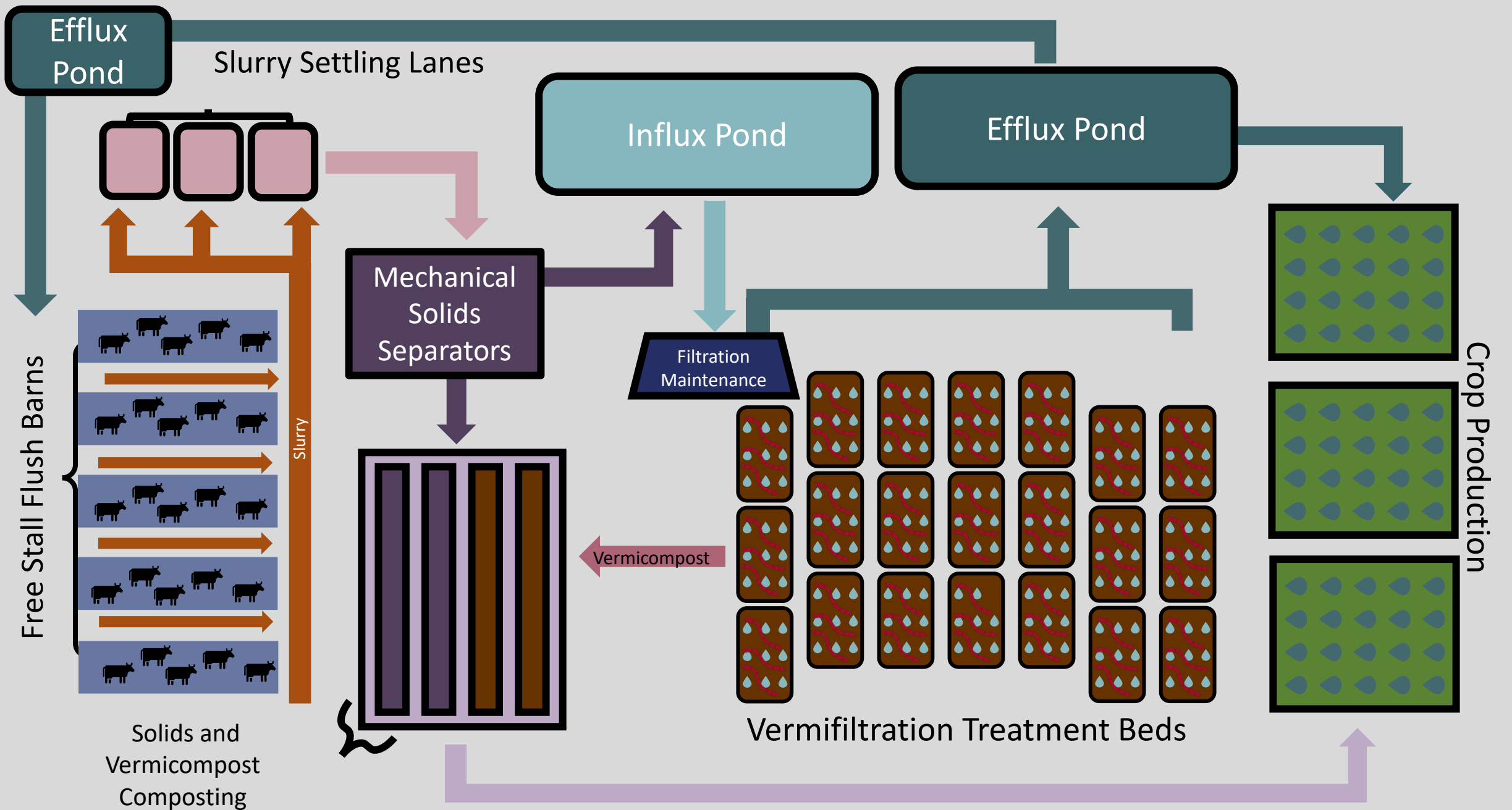
CARBON BANK
Soil acts much like a bank, storing carbon and filling its vault as carbon deposits are made from plants and animals



Source: Australian Soil Management

How to create a soil management bank to assist with carbon grazing.

VERMIFILTRATION SYSTEM



VERMIFILTRATION EFFECTS ON LIQUID CATTLE MANURE

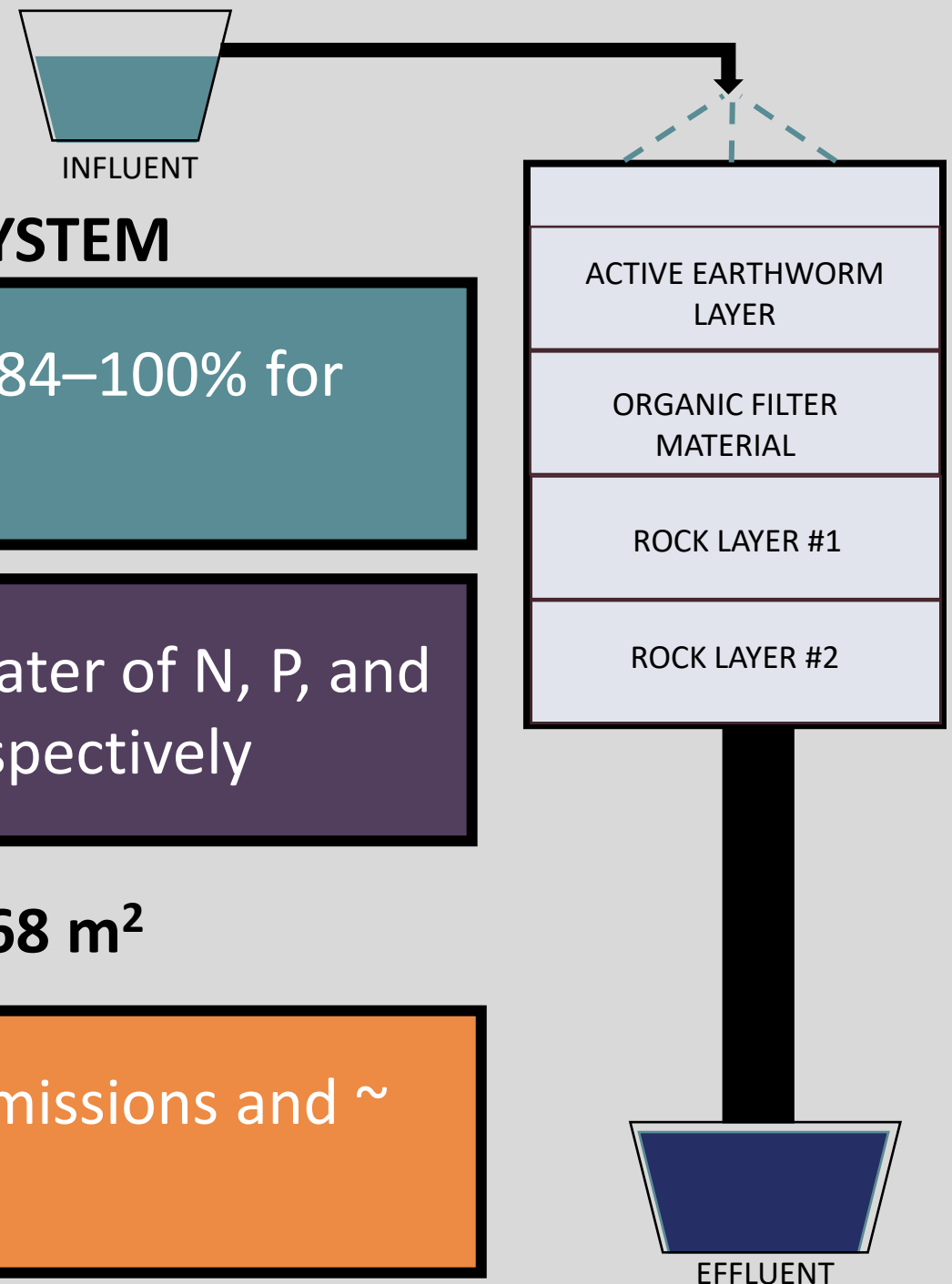
SHIPPING CONTAINER VERMIFILTRATION SYSTEM

Miito et al. (2022)--emission reductions of 84–100% for NH_3 , 95–100% for CH_4

Miito et al. (2021)--reductions in effluent water of N, P, and suspended solids by 77%, 48% and 68%, respectively

SMALL SCALE VERMIFILTRATION SYSTEM ~68 m²

Dore et al. (2022)-- 97% reduction in CH_4 emissions and ~83.5% reduction in N and P in effluent



QUESTIONS TO ADDRESS

How does ambient temperature affect gas fluxes?

What is the variability across a vermifiltration system?

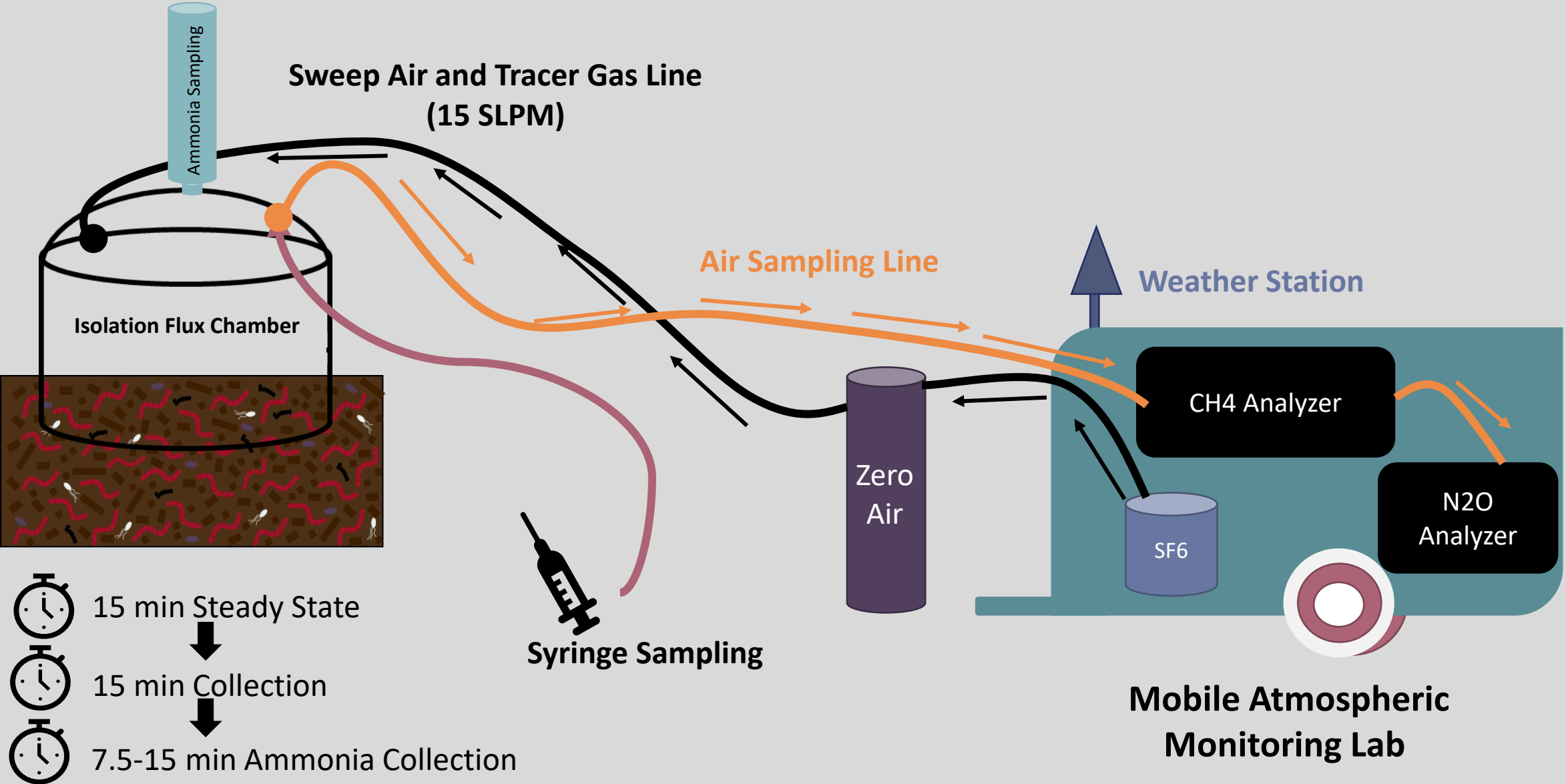
How does the process affect the chemical and physical components of liquid manure?

VERMIFILTRATION BED SAMPLING

Open Crop Field



GHG AND AMMONIA MEASUREMENTS



GHG AND AMMONIA EMISSION CALCULATIONS

$$E_{gas} = Q_s \left(\frac{C_T}{C_{Chamber\ T}} \right) C_{Chamber\ gas} - Q_s C_{s(gas)} - \left[\left\{ Q_s \left(\frac{C_T}{C_{Chamber\ T}} \right) - Q_s \right\} C_{A(gas)} \right]$$

E_{gas} = emission rate of gas

Q_s = Zero air flow rate

C_T = Expected concentration of SF₆ 3200 ppt

$C_{Chamber\ T}$ = measured SF₆ from syringe sample

$C_{Chamber\ gas}$ = measured gas (N₂O, CH₄, NH₃) from syringe

$C_{s(gas)}$ = measured gas (N₂O, CH₄, NH₃) in zero air

$C_{A(gas)}$ = measured gas (N₂O, CH₄, NH₃) in ambient air

STATISTICAL ANALYSIS

SAS: Proc Mixed

Split Plot Design

Evaluated effect of temperature, bed, and bed by ambient temperature

ENVIRONMENTAL CONDITIONS

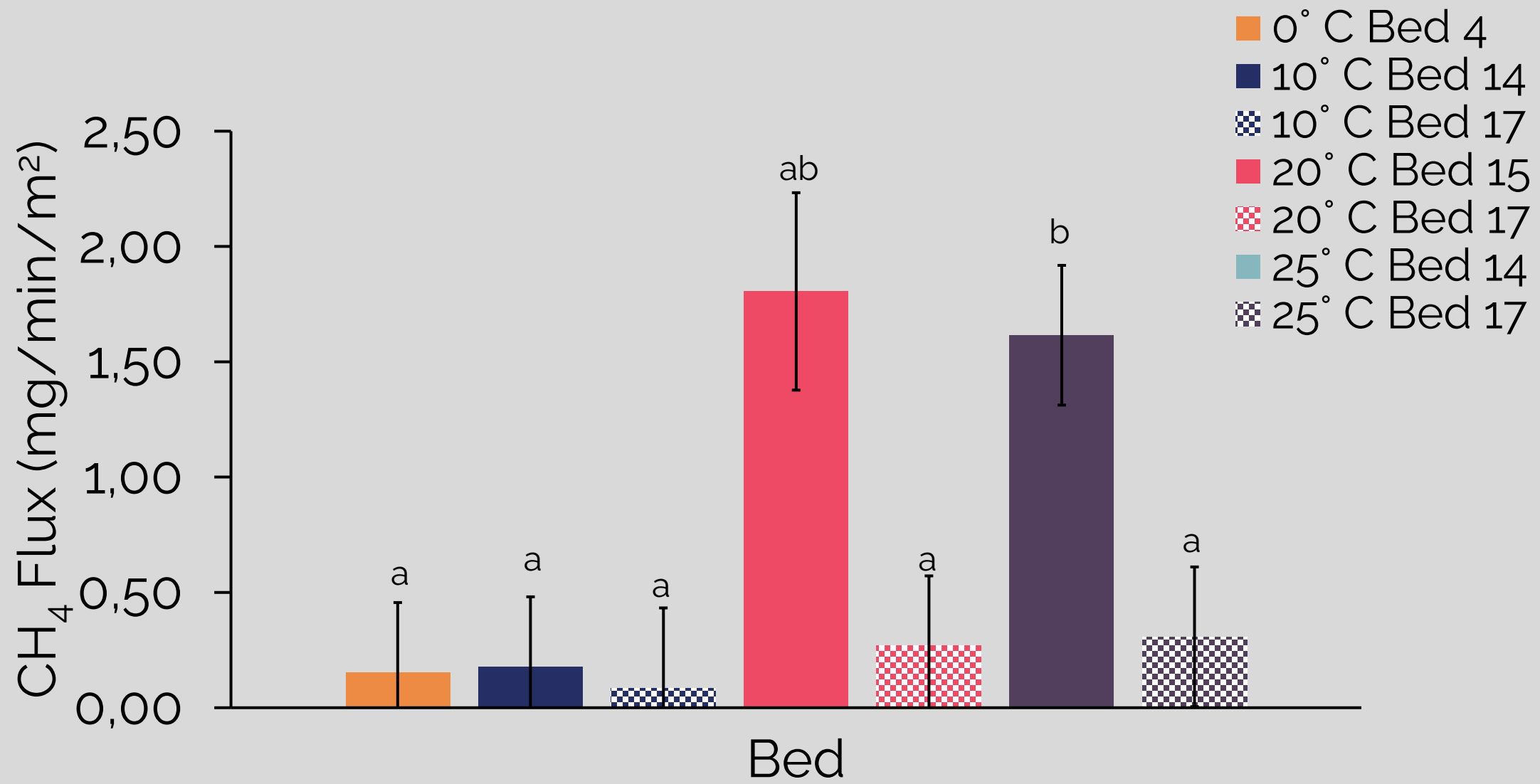
0 °C			
	Min	Max	Average
Windspeed, m/s	0.3	3.8	1.8
Temperature, °C	0.8	2.7	2.0
Relative Humidity, %	53.0	65.3	58.1
Pressure, hPa	983.9	984.6	984.1

20 °C			
	Min	Max	Average
Windspeed, m/s	0.1	5.8	2.3
Temperature, °C	20.2	25.2	22.5
Relative Humidity, %	38.4	60.0	48.5
Pressure, hPa	957.8	959.5	958.9

10 °C			
	Min	Max	Average
Windspeed, m/s	0.1	4.8	1.6
Temperature, °C	4.6	13.6	9.9
Relative Humidity, %	32.3	58.0	42.1
Pressure, hPa	978.0	981.2	979.8

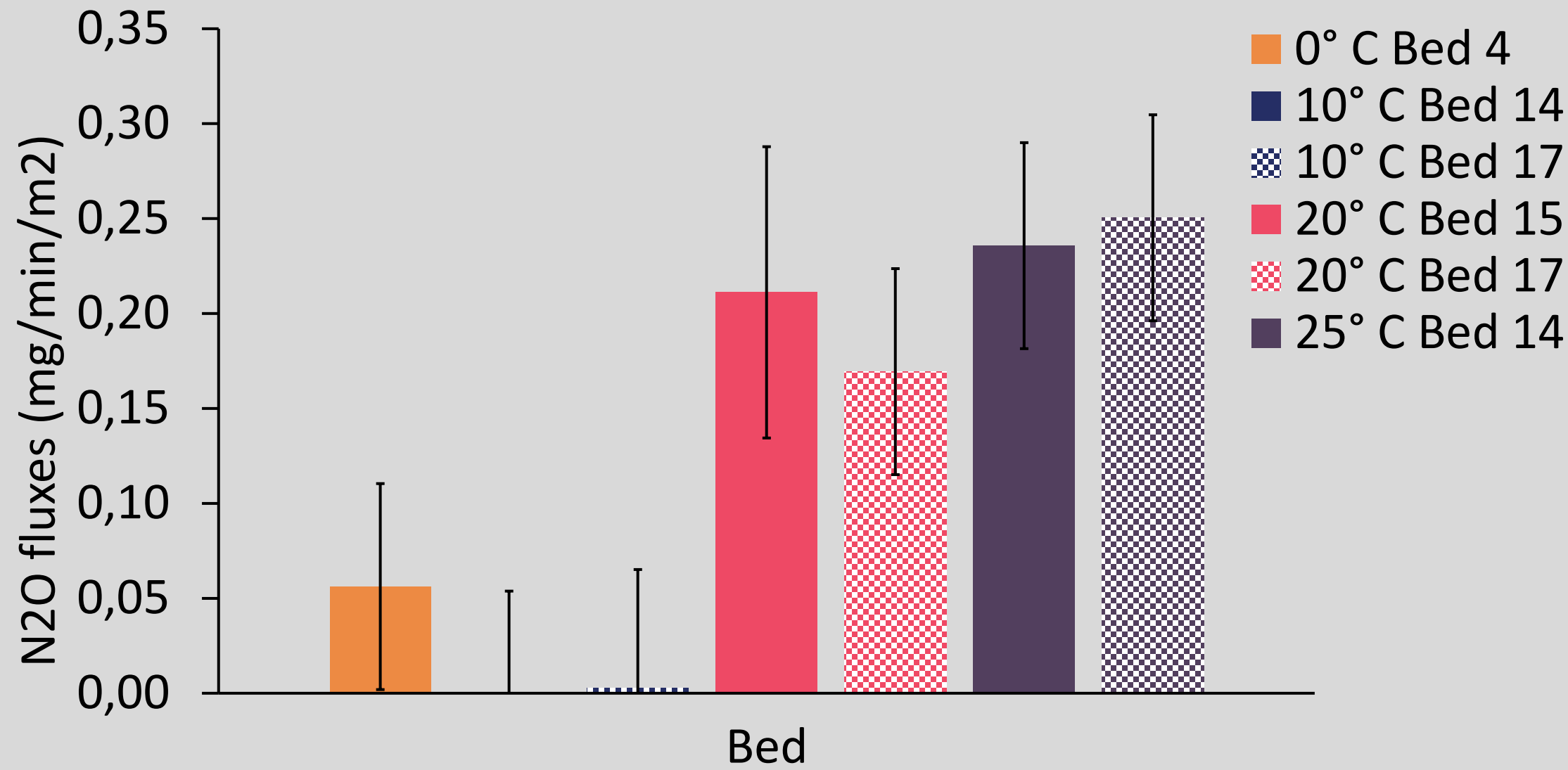
25 °C			
	Min	Max	Average
Windspeed, m/s	0.3	8.2	2.6
Temperature, °C	19.6	25.6	23.5
Relative Humidity, %	34.2	63.1	43.6
Pressure, hPa	968.1	969.1	968.8

CH₄ FLUXES BY AMBIENT TEMPERATURE & BED



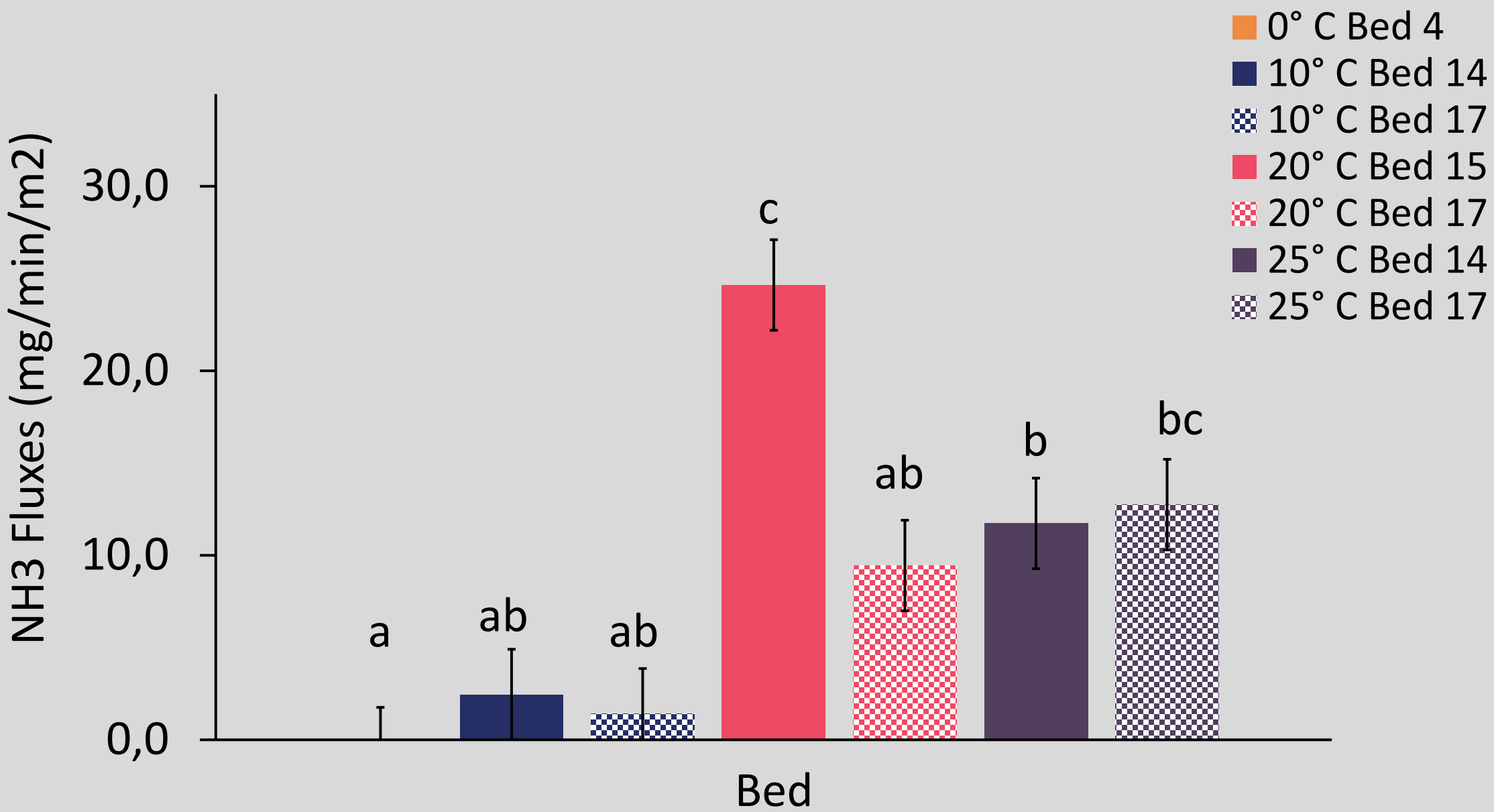
P-Value <0.05

N₂O FLUXES BY AMBIENT TEMPERATURE & BED



P-Value <0.05

NH₃ FLUXES BY AMBIENT TEMPERATURE & BED



P-Value <0.05

SUMMARY OF GHG and NH₃ EMISSIONS

ESTIMATED YEARLY EMISSIONS FROM 3 HA WORM BEDS

Ambient Temperature	Days in Temperature Period	NH ₃ metric tons	CH ₄ metric tons CO ₂ -eq	N ₂ O metric tons CO ₂ -eq
0° C	57	0.06	10.23	36.60
10° C	113	9.14	18.16	1.05
20° C	107	76.37	97.96	224.21
25° C	88	44.46	99.19	244.53
TOTAL	365	130.03	225.55	506.39

ESTIMATED YEARLY EMISSIONS FROM LAGOON SYSTEM SCALED TO 3 HA

Ambient Temperature	Days in Temperature Period	NH ₃ metric tons	CH ₄ metric tons CO ₂ -eq	N ₂ O metric tons CO ₂ -eq
0° C	57	3.50	902.18	384.61
10° C	113	7.18	15,030.96	444.03
20° C	107	5.10	5,685.30	101.93
25° C	88	4.20	4,675.76	83.83
TOTAL	365	19.98	26,294.20	1,014.39

(LEYTEM ET AL., 2011)

85%

100%

55%

NITROGEN AND PHOSPHOROUS COMPOSITION IN INFLUENT/EFFLUENT

NITROGEN									
Sample ID	TEMPERATURE (°C)	Organic N ppm	%	Ammonium (NH ₄ ⁺) ppm	%	Nitrate (NO ₃ ⁻) ppm	%	Total N ppm	%
INFLUENT	0	141.7 ^a	54%	498.0 ^a	75%	0.6 ^a	83%	640.3 ^a	71%
EFFLUENT	0	65.0 ^a		122.4 ^b		1.1 ^a		188.5 ^b	
INFLUENT	10	552.8 ^a	79%	931.6 ^a	90%	0.2 ^a	50%	1,484.5 ^a	86%
EFFLUENT	10	114.4 ^a		97.5 ^b		0.1 ^a		212.1 ^b	
INFLUENT	20	532.3 ^a	64%	657.1 ^a	61%	1.2 ^a	83%	1,190.6 ^a	62%
EFFLUENT	20	193.9 ^a		258.8 ^b		0.2 ^a		452.9 ^b	
INFLUENT	25	722.5 ^a	41%	644.2 ^a	72%	0.3 ^a	0%	1,367.0 ^a	56%
EFFLUENT	25	422.8 ^a		180.9 ^b		0.3 ^a		604.0 ^b	

PHOSPHOROUS					
Sample ID	TEMPERATURE (°C)	Phosphate (P ₂ O ₅) ppm	%	Orthophosphate (PO ₄ ⁻ P) (ppm)	%
INFLUENT	0	242.6 ^a	72%	44.1 ^a	68%
EFFLUENT	0	69.0 ^b		14.1 ^b	
INFLUENT	10	569.6 ^a	87%	84.8 ^a	83%
EFFLUENT	10	72.2 ^b		14.3 ^b	
INFLUENT	20	522.5 ^a	77%	73.0 ^a	67%
EFFLUENT	20	120.0 ^b		24.3 ^b	
INFLUENT	25	505.1 ^a	51%	21.3 ^a	21%
EFFLUENT	25	247.9 ^b		25.8 ^b	

CHEMICAL AND PHYSICAL CHANGES IN INFLUENT AND EFFLUENT

Potassium (K_2O), Sulfur (S), Calcium (Ca), Magnesium (Mg), Sodium (Na), Zinc (Zn), Iron (Fe), Manganese (Mn), Copper (Cu), Boron (B), Soluble Salts, pH, Dry Matter, Organic Matter, Ash, Total Suspended Solids (TSS), Total Dissolved Solids (TDS), Volatile Suspended Solids (VSS), Total Solids, Total Carbon Organic Carbon, C: N



Ca, Mg, Zn, Fe, Mn, TSS, and Total C at 0 °C, 10 °C, 20 °C, and 25 °C

P value <0.05

MASS BALANCE



P	100%	=	0%	+	63%	+	29%	8%
K	100%	=	0%	+	9%	+	105%	-5%
C	100%	=	1%	+	53%	+	39%	7%
N	100%	=	26%	+	19%	+	30%	25%

CONCLUSION

Vermifiltration is a “low-cost” technology to treat liquid dairy waste that reduced CH_4 and N_2O emissions, but increases NH_3 emission compared to literature values

Vermifiltration alters wastewater components

Dairies may participate in carbon markets and vermicompost sales

FUTURE WORK

Implement technologies to mitigate NH_3 e.g. lower pH of influent or addition of biochar

Create a whole farm emission profile with diet/management changes

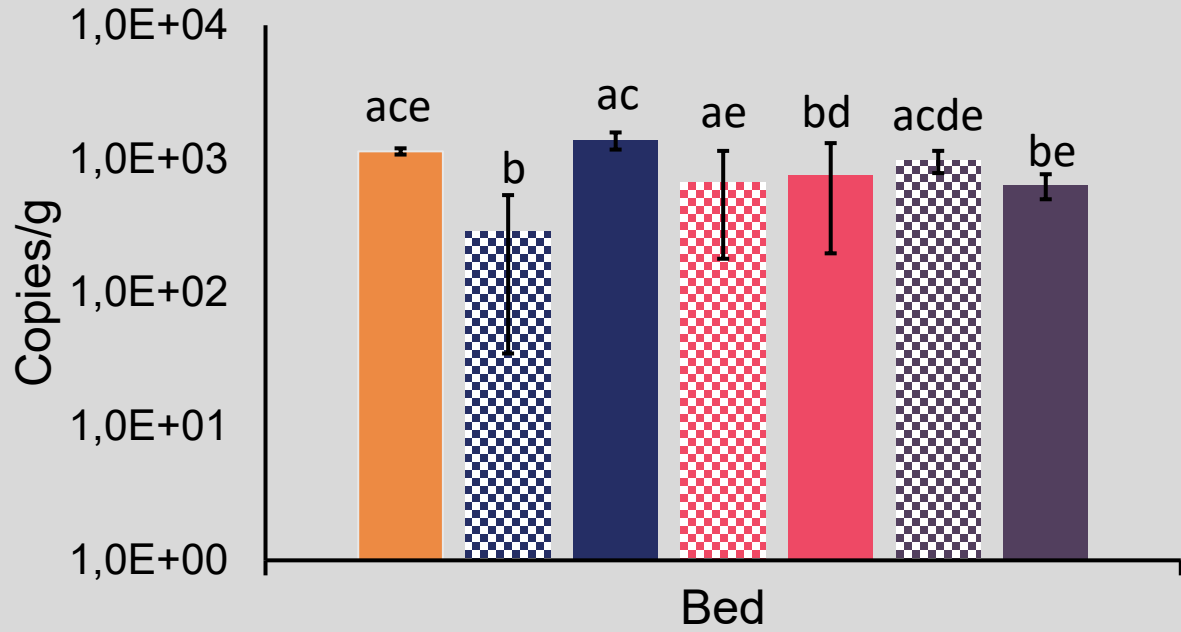
The whole farming enterprise must be examined when considering mitigation strategies so reduction in one species does not inadvertently result in higher emissions of another pollutant.



CHEMICAL AND PHYSICAL PROPERTIES IN VERMIFILTRATION BED MATERIAL AND VERMICOMPOST

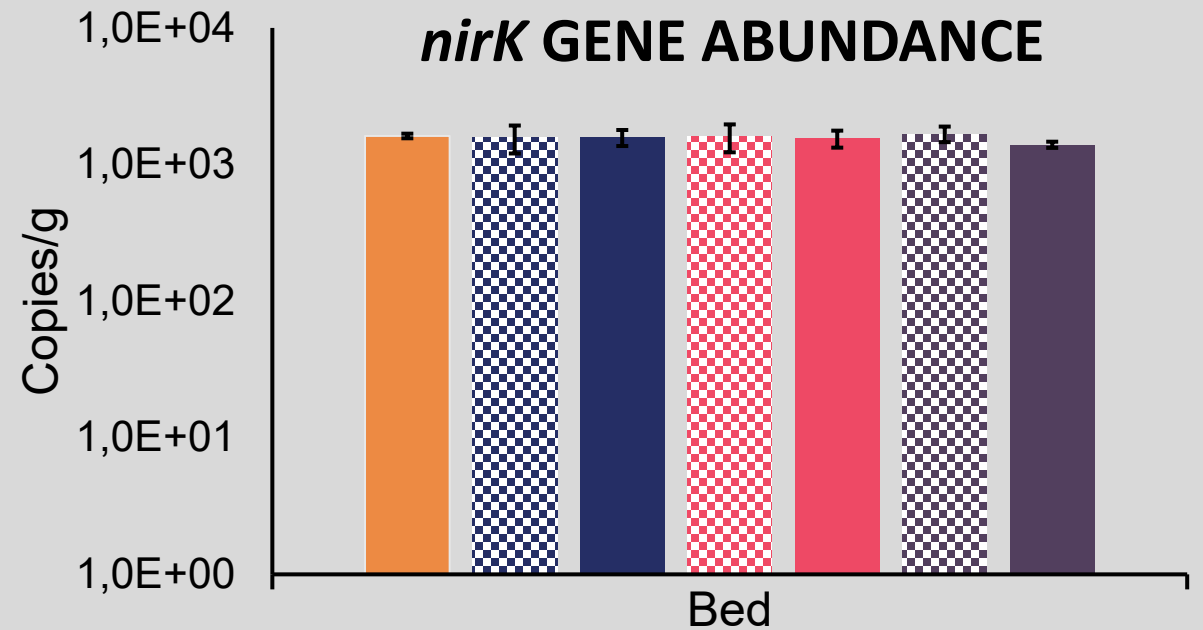
Item	BED MATERIAL		VERMIPOST	
	Average	SE	Average	SE
Organic N %	1.86 ^a	0.09	3.14 ^b	0.33
Ammonium (NH ₄ ⁺) %	0.02 ^a	0.01	0.01 ^a	0.02
Nitrate (NO ₃ ⁻) %	0.01 ^a	0.00	0.17 ^b	0.00
Total N %	1.88 ^a	0.09	3.32 ^b	0.33
Phosphate (P ₂ O ₅) %	2.67 ^a	0.15	4.52 ^b	0.55
Orthophosphate (PO ₄ -P) %	0.11 ^a	0.01	0.17 ^a	0.05
Potassium (K ₂ O) %	1.76 ^a	0.20	2.36 ^a	0.73
Sulfur (S) %	0.49 ^a	0.03	1.00 ^b	0.10
Calcium (Ca) %	6.04 ^a	0.54	7.08 ^a	1.96
Magnesium (Mg) %	1.69 ^a	0.10	1.39 ^a	0.35
Molybdenum (Mo) %	2.75 ^a	0.54	3.52 ^a	1.94
Sodium (Na) %	0.69 ^a	0.10	0.45 ^a	0.35
Zinc (Zn) ppm	276.72 ^a	14.93	496.80 ^b	53.87
Iron (Fe) ppm	5880.80 ^a	513.24	5556.70 ^a	1850.53
Manganese (Mn) ppm	334.84 ^a	14.81	668.70 ^b	53.39
Copper (Cu) ppm	48.78 ^a	1.95	76.10 ^b	7.03
Boron (B) ppm	53.62 ^a	2.31	72.10 ^a	8.33
Soluble Salts mnho/cm	23.81 ^a	3.80	49.66 ^a	13.69
pH	9.24 ^a	0.17	8.10 ^a	0.63
Ash %	35.22 ^a	1.93	39.40 ^a	6.97
Organic Matter %	64.78 ^a	1.93	60.60 ^a	6.97
Moisture %	73.37 ^a	0.83	58.28 ^b	2.98
Dry Matter %	26.63 ^a	0.83	41.72 ^b	2.98
Organic Carbon %	37.57 ^a	1.12	35.15 ^a	4.04
C: N Ratio	22.12 ^a	1.12	11.20 ^b	4.04
Total Carbon %	31.46 ^a	1.09	32.51 ^a	3.94

***nirS* GENE ABUNDANCE**

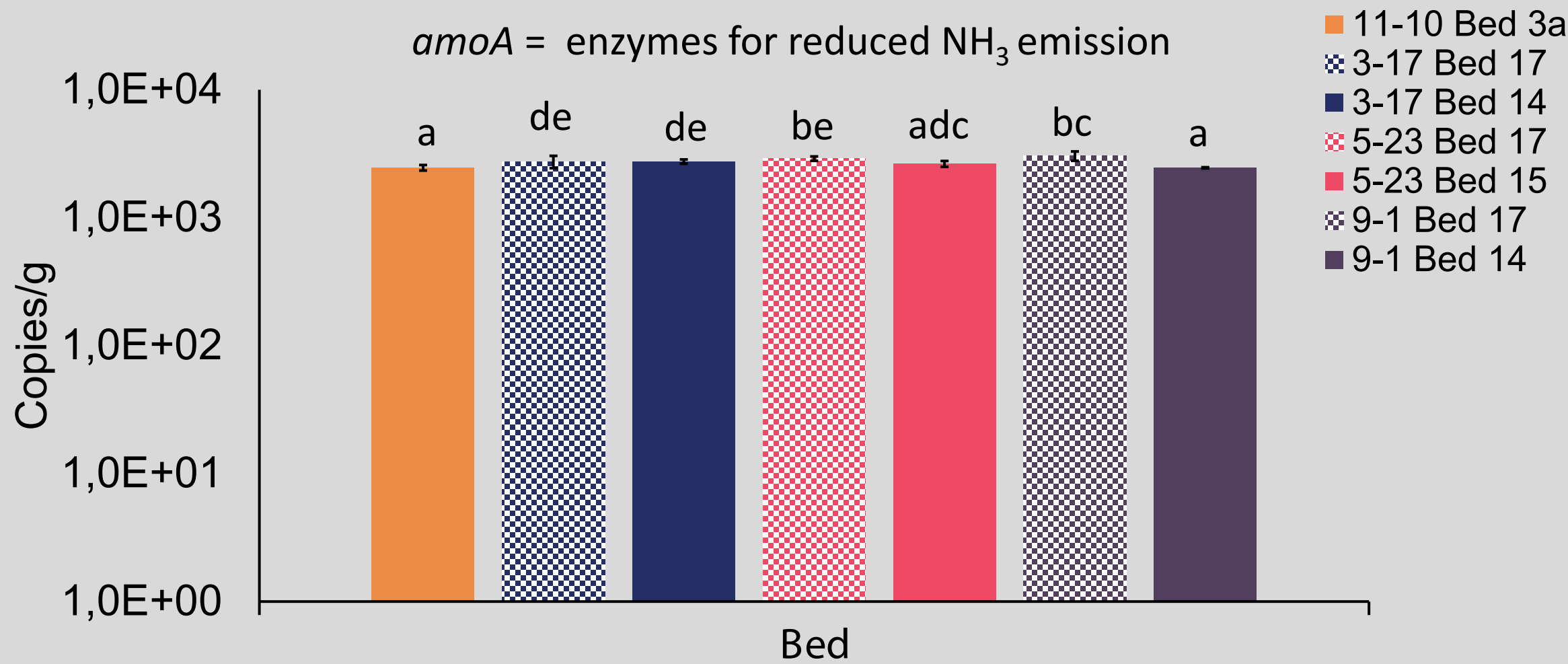


Nir genes = enzymes for nitrite to NO gas

***nirK* GENE ABUNDANCE**



amoA GENE ABUNDANCE IN WORM BEDS



(Falcon et al., unpublished)

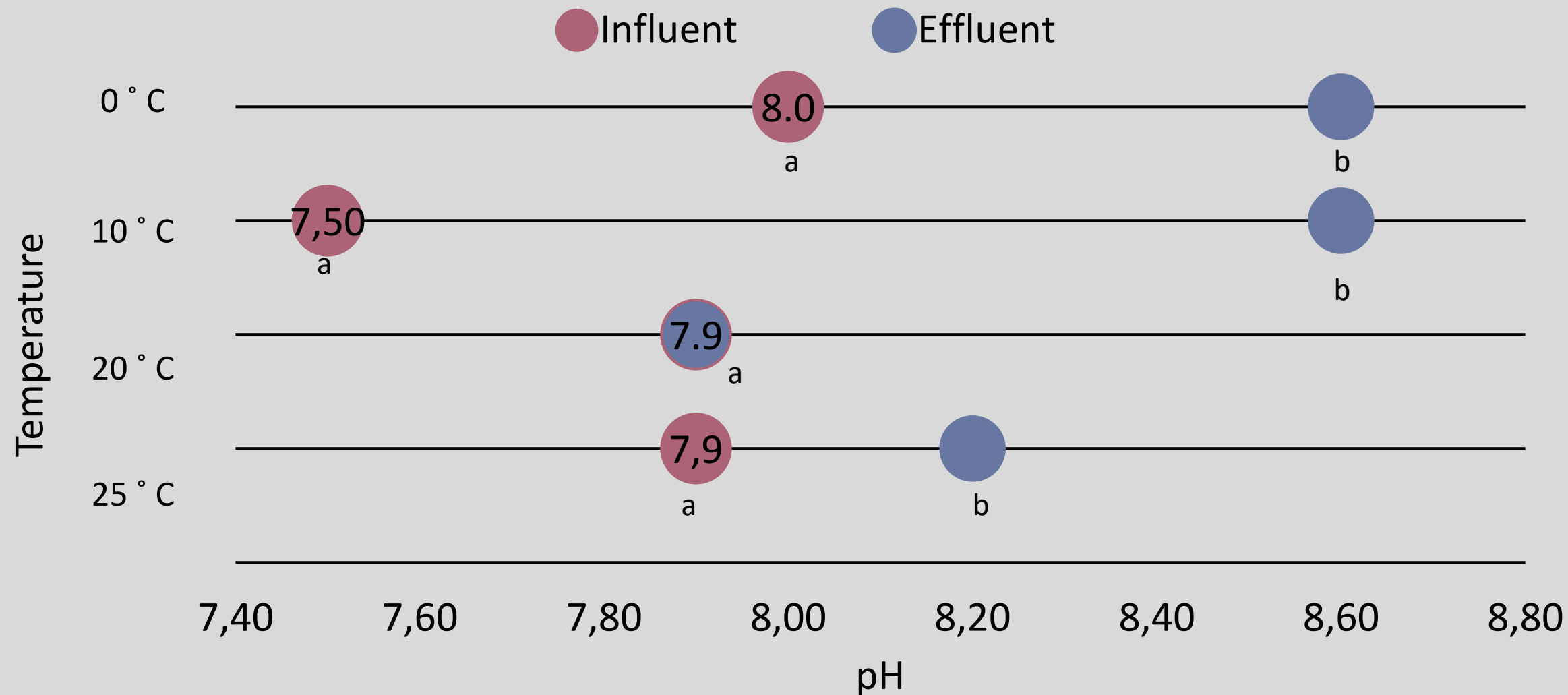
P-Value <0.05

nosZ GENE ABUNDANCE IN WORM BEDS



(Falcon et al., unpublished)

VARIATION IN pH IN INFLUENT AND EFFLUENT WATER



P value <0.05

SOLID AND LIQUID CHEMICAL COMPOSITION & ANALYSIS

